## SYNTHESES OF DICTYOPTERENE B (HORMOSIRENE) AND ITS ENANTIOMER VIA ASYMMETRIC S<sub>CN</sub>, REACTIONS<sup>1a</sup>

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Abstract: Stereocontrolled syntheses of dictyopterene B (hormosirene) (<u>1</u>) and its enantiomer (ent-<u>1</u>) are reported. Key steps are highly stereoselective  $S_{cN}$ , reactions of esters of alcohols <u>2</u> and <u>5</u> derived from (+)-camphor.

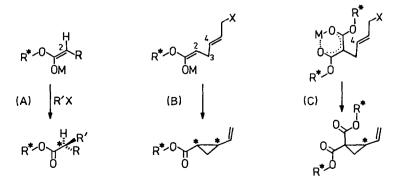
Dictyopterene B (1), the main constituent of the essential oils from several Dictyopteris spe-



cies, was first isolated by Moore in 1970<sup>3</sup>. Even at a low level of concentration, this compound displays an intense "ocean smell". More recently, Müller, Jaenicke and their co-workers have detected 1, named hormosirene by them, in a variety of other brown algae, and have shown that it acts as sperm attractant<sup>4</sup>.

As part of a programme aimed at developing general asymmetric syntheses for 2-vinylcycloalkane carboxylic acids, useful starting materials for natural product syntheses, we have synthesized both dictyopterene B (1) and its enantiomer (ent-1) in high enantiomeric and diastereomeric purity. As we learned after completion of this work, syntheses via a different route have simultaneously been achieved by Jaenicke's group<sup>5</sup>. Previously reported syntheses, usually involving separations of stereoisomers, have yielded racemic $^{6}$  and optically active material of 85% ee<sup>7</sup>.

Key step of our sequence is an  $S_{cN}$ , reaction of type B, using alcohols 2 and 5 (Scheme 1) as chiral reagents R\*-OH. With these high levels of diastereoselection can be obtained in ester alkylations via lithium enolates according to A, a reaction involving diastereoface selection at center C-2<sup>8</sup>. On the other hand, Quinkert and co-workers<sup>9</sup> in their highly stereoselective  $S_{cN}$ , reaction C (R\*-OH = 8-phenylmenthol) have demonstrated the feasibility of effective stereocontrol at the remote center C-4. In view of these examples it was of interest to examine reaction B which involves both centers, C-2 and C-4, and thus allows simultaneous construction of both stereogenic centers of the three-membered ring.



Ar =

Scheme 1

a, b С .N-S0₂Ph S02Ph ∖v-S0₂Ph Аr 89% 'nн 2 3 trans cis вг <u>a</u> (1R, 2S) (1R, 2R) <u>도</u>, <u>d</u> <u>b</u> (15, 2R) (15, 25) a, b с SO\_P SO<sub>2</sub>Ph 77% ń 5 <u>6</u> <u>7</u>

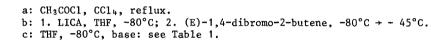
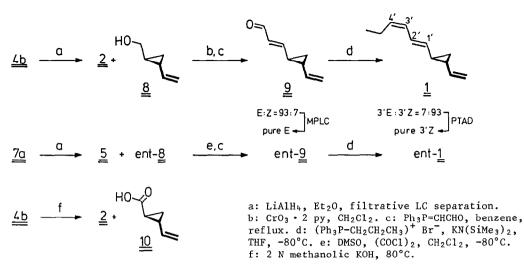


Table 1. S <sub>cN'</sub>	Reactions	of	Esters	3	and	<u>6</u>
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Entry	Ester	Base	<u>4/7</u>							Yield [%] <sup>b)</sup>
			<u>a</u>	:	Þ	:	<u>C</u>	:	<u>d</u> a)	
1	<u>3</u>	1.2 eq LICA	48	:	48	:	2.5	:	1.5	38 (42)
2	<u>3</u>	1.2 eq LICA/HMPT	33	:	58	:	5	:	4	12
3	<u>3</u>	1.3 eq KO <sup>t</sup> Bu	14	:	75	:	3	:	12	87 (100)
4	<u>3</u>	<u>3.6 eq KO<sup>t</sup>Bu + 1.8 eq H₂O<sup>c)</sup> 3 cryst.</u>			96.5 99.5		0.5			<u>85 (100)</u> 80 (92)
5	<u>3</u>	7 eq KO <sup>t</sup> Bu + 4 eq H <sub>2</sub> O <sup>d)</sup> 5 cryst.	5 0.1	:	92 99.5		1 0.4	-	2 0	100 88
6	<u>6</u>	2 eq KO <sup>t</sup> Bu	74	:	23	:	1.5	:	1.5	82
7	<u>6</u>	4 eq KO <sup>t</sup> Bu + 2 eq H <sub>2</sub> O <sup>C)</sup> 5 cryst.	83 99.5		11.4 0.5	: :			2.8 0	88 59
8	<u>6</u>	<u>7 eq KO<sup>t</sup>Bu + 4 eq H<sub>2</sub>O<sup>d</sup>) 3 cryst.</u>	92.5 99.5		7	:	0.2 0		0.3	<u>100</u> <u>74</u>

a) Analysis by HPLC. b) Values in brackets: yields corrected with respect to recovered starting material. c) Solid KO Bu reacted with gaseous  $H_2O$ . d) KO Bu in THF reacted with a solution (7%) of  $H_2O$  in THF, -80°C. Scheme 2



The requisite intermediates  $\underline{3}$  and  $\underline{6}$  were prepared from  $\underline{2}$  and  $\underline{5}$  via straightforward 2-step sequences in 89 and 77% yield, respectively (Scheme 1). Their four diastereomeric cyclization products  $\underline{4a} - \underline{d}$  and  $\underline{7a} - \underline{d}$ , respectively, were found to be separable by liquid chromatography (eluent: petroleum ether/ethyl acetate 95:5, silica gel) which allowed convenient product analysis (HPLC) and isolation of the individual (crystalline) isomers (MPLC). Assignment of relative configuration for the cyclopropane moiety is based on <sup>1</sup>H NMR coupling constants<sup>10</sup>; absolute configuration was only established for the trans isomers, by transformation of  $\underline{4b}$  and  $\underline{7a}$  into 1 and ent-1, respectively, of known configuration<sup>3</sup>.

Stereoselectivity of the cyclization reaction was found to be highly dependent on the base system applied (Table 1). Thus, in contrast to previous successful applications in alkylations of type A, lithium cyclohexylisopropylamide (LICA) failed to give satisfactory results upon reaction with  $\underline{3}$  (entries 1,2). Of other bases then investigated for the cyclization of  $\underline{3}$ , K0<sup>t</sup>Bu (THF) proved most successful (entries 3-5), but showed properties dependent on the degree of purity. Thus, a partially decomposed sample gave a higher level of diastereoselection than the freshly sublimed pure compound. Following this lead, we then used various mixtures obtained by controlled addition of water to pure K0<sup>t</sup>Bu<sup>11</sup>. Finally, conditions as specified by entry 4 of Table 1 produced the isomer  $\underline{4b}$  with excellent diastereoselectivity<sup>12</sup>. As anticipated on the basis of prior experience<sup>8</sup>, the exo isomer <u>6</u> behaved similarly to <u>3</u> and, also expected, preferentially gave the isomer  $\underline{7a}$  with absolute configuration of the cyclopropane moiety opposite to that of  $\underline{4b}$ .

Cyclizations on a 10 g scale followed by recrystallization afforded the cyclopropane derivatives <u>4b</u> and <u>7a</u> 99.5% pure<sup>13</sup> in 74 - 92% yield. Saponification of <u>4b</u> (Scheme 2) gave the known<sup>7</sup> carboxylic acid (+)-(1S,2R)-<u>10</u>,  $[\alpha]_{D}^{26} = +194$  (c = 1.3, 95% EtOH),  $[\alpha]_{D}^{26} = +178$  (c = 1.1, EtOH)<sup>7</sup>, in 74% overall yield from <u>2</u>. Alternatively, reductive ester cleavage of <u>4b</u> and <u>7a</u> followed by filtrative LC separation of products gave the alcohols <u>8</u> and ent-<u>8</u> in overall yields of 87 and 49% from the reagents <u>2</u> and <u>5</u>, respectively; <u>8</u>:  $\alpha_{579}^{279} = +61.2$  (neat, d = 1 dm)<sup>14</sup>,  $[\alpha]_{D}^{26} = +60.1$  (c = 1.0, 95% EtOH),  $[\alpha]_{D}^{26} = +54$  (c = 1, EtOH)<sup>7</sup>; ent-<u>8</u>:  $\alpha_{579}^{279} = -60.5$ 

(neat, d = 1 dm)<sup>14</sup>; as the reductive ester cleavage cannot cause racemization, enantiomeric purities of <u>8</u> and ent-<u>8</u> must be identical to the diastereometric purities of the precursors  $^{13}$ . i. e. 99.8 and 99.0% (Table 1), respectively.

For transformation of the alcohols  $\underline{8}$  and ent- $\underline{8}$  into the target compounds stereoselective Wittig olefination methods worked out by Bestmann and co-workers<sup>15</sup> (Scheme 2) were used. The requisite aldehydes were obtained by Collins<sup>16</sup> or Swern<sup>17</sup> oxidation. Their reaction with the phosphorane  $Ph_3P=CHCHO$  afforded the enals <u>9</u> and ent-<u>9</u> as 93:7 E/Z mixtures (HPLC analysis). MPLC purification gave the pure E isomers in 52 and 32% yield from 8 and ent-8, respectively; <u>9</u>:  $\alpha_{379}^{2}$  = +230 (neat, d = 1 dm)<sup>14</sup>. Finally, reaction with the "salt-free" phosphorane Ph<sub>3</sub>P=CHCH<sub>2</sub>CH<sub>3</sub> yielded <u>1</u> and ent-<u>1</u> as 93: 7 3'Z/3'E mixtures (GLC analysis<sup>18</sup>). The 3'E isomers were selectively removed by reaction with 4-phenyl-1,2,4-triazolidine-3,5-dione (PTAD)  $(CH_2Cl_2, -80^{\circ}C)^{19}$  to give pure dictyopterene B (<u>1</u>) and its enantiomer in yields of 59 and 50%, respectively; 1:  $[\alpha]_{\Omega}^{24} = -41.6$  (c = 0.53, CHCl<sub>3</sub>),  $[\alpha]_{\Omega}^{24} = -43$  (c = 10, CHCl<sub>3</sub>)<sup>3</sup>; ent-1:  $[\alpha]_{\Omega}^{24} = -43$ + 41.3 (c = 0.67, CHCl<sub>3</sub>).

According to evaluation by experts (BASF AG), olfactory properties of 1 and ent-1 differ considerably. In particular, only <u>1</u> displays the characteristic very intense, rather pleasant, "ocean smell". The odour of ent-1 is less intense and devoid of this note.

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   New address: Organisch-Chemisches Institut der Universität Heidelberg, Im Neuenheimer Feld
- 270, D-6900 Heidelberg.
- 3. J.A. Pettus Jr., R.E. Moore, J. Chem. Soc., Chem. Commun. 1970, 1093; R.E. Moore, J.A. Pettus Jr., J. Mistysyn, J. Org. Chem. <u>93</u>, 2201 (1974). 4. D.G. Müller, M.N. Clayton, G. Gassmann, W. Boland, F.-J. Marner, L. Jaenicke, Experientia
- 40, 211 (1984); D.G. Müller, M.N. Clayton, G. Gassmann, W. Boland, F.-J. Marner, T. Schot-Ten, L. Jaenicke, Naturwiss. <u>72</u>, 97 (1985).
- 5. T. Schotten, W. Boland, L. Jaenicke, Helv. Chim. Acta, in press. We thank Prof. Jaenicke for a preprint.
- A. Ali, D. Sarantakis, B. Weinstein, J. Chem. Soc., Chem. Commun. <u>1971</u>, 940; M.P. Schneider, M. Goldbach, J. Am. Chem. Soc. <u>102</u>, 6114 (1980); T. Akintobi, L. Jaenicke, F.-J. Marner, S. Waffenschmidt, Liebigs Ann. Chem. <u>1979</u>, 986; W. Boland, P. Ney, L. Jaenicke, Syn-
- ner, S. Wattenschmidt, Liebigs Ann. Chem. 1979, 900; W. Doland, F. Ney, L. Gaenicke, Synthesis 1980, 1015.
  7. T. Kajiwara, T. Nakatomi, Y. Sasaki, A. Hatanaka, Agric. Biol. Chem. 44, 2099 (1980).
  8. R. Schmierer, G. Grotemeier, G. Helmchen, A. Selim, Angew. Chem. 93, 209 (1981), Int. Ed. Engl. 20, 207 (1981); G. Helmchen, R. Schmierer, Tetrahedron Lett. 1983, 1235; G. Helmchen, A. Selim, D. Dorsch, I. Taufer, Tetrahedron Lett. 1983, 3213.
  9. G. Quinkert et al., Liebigs Ann. Chem. 1982, 1999.
  10. Trans isomers <u>4a,b</u> and <u>7a,b</u>: J<sub>1,2</sub> = 4.1 ± 0.1 Hz; cis isomers <u>4c,d</u> and <u>7c,d</u>: J<sub>1,2</sub> = 8.5 Hz.
  11. KOH does not effect cyclization of <u>3</u>.
  12. Control experiments showed that under the reaction conditions K0<sup>t</sup>Bu does not induce epimerization at C-1: i.e. the reaction is kinetically controlled.

- rization at C-1; i. e. the reaction is kinetically controlled.
- 13. If desired, <u>4b</u> and <u>7a</u> may be obtained pure by additional recrystallization steps and/or MPLC.

- 14. Not normalized with respect to density.
  15. H.J. Bestmann et al., Liebigs Ann. Chem. 1982, 1359.
  16. J.C. Collins, W.W. Hess, F. J. Frank, Tetrahedron Lett. 1968, 3363.
  17. A.J. Mancuso, S.-L. Huang, D. Swern, J. Org. Chem. 43, 2480 (1978).
  18. GLC: quartz capillary column 25 m × 0.32 mm, coated with DB 5, inj. temp. 200°C, column temp. 50°C.
- 19. W. Boland, K. Mertes, Helv. Chim. Acta 67, 616 (1984).

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